

**Full Length Research Paper****Structure and Room Temperature Ferromagnetism of Ni<sup>2+</sup>-doped ZnO Nanoparticles Prepared by Sol-gel Process**Trilok Kumar Pathak<sup>1\*</sup>, L.P.Purohit<sup>2</sup><sup>1,2</sup> Department of Physics, Gurukul Kangri Vishwavidyalaya, Haridwar (U.K.)-India*\*Corresponding Author: Trilok Kumar Pathak***Abstract**

Ni doped ZnO (Zn<sub>1-x</sub>Ni<sub>x</sub>O, 0 ≤ x ≤ 0.05) Diluted magnetic semiconductors (DMSs) nanoparticles are prepared by sol-gel process. Transmission electron microscopy (TEM) image shows nanoparticles with an average size of 25nm. The analysis of X-ray diffraction (XRD) identified that the impurity phase is observed when the Ni content x reaches 0.05. With the increment of x, wurtzite structures degrade gradually. The magnetic properties are measured using superconducting Quantum interference device (SQUID) at room temperature; the Zn<sub>1-x</sub>Ni<sub>x</sub>O (x ≤ 0.02) nanoparticle shows ferromagnetism, however, for the sample of Zn<sub>1.05</sub>Ni<sub>0.05</sub>O paramagnetism is observed.

**Keywords :** ZnNiO nanoparticle, Structure and ferromagnetism**Introduction**

ZnO has been extensively investigated due to its potentials in optoelectronic applications. It has wide band gap (3.37 eV) and a large exciton binding energy (60 meV) [1-4]. Recent predictions of ferromagnetism at room temperature in III-V and II-VI semiconductor materials have attracted renewed interest from the scientific and industrial communities [5-7]. The incorporation of ions such as Co, Ni, Fe and Cu into the structure of ZnO could open new possibilities involving magnetic, magneto-optic and magneto transport effects [8-10]. Among them most studies focused on Co-doped ZnO [11,12] while fewer studies on Ni-doped ZnO have been reported.

Transition metal (TM) doped ZnO can be prepared by a variety of methods such as chemical vapor deposition (CVD), Pulse laser deposition (PLD) and sol-gel process. The sol-gel process has been successfully used to obtain nanoscale material at low cost. In this work, the sol-gel process is developed to prepare ZnNiO nanoparticle, the structure and magnetism are studied in details.

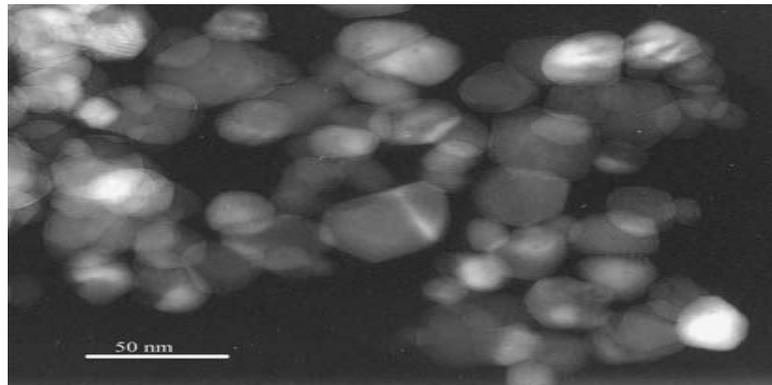
**Experimental Details**

Zn<sub>1-x</sub>Ni<sub>x</sub>O nanoparticles are fabricated by sol-gel process. First, at room temperature, tetramethyl ammonium hydroxide (N(Me)<sub>4</sub>OH · 5H<sub>2</sub>O) was dissolved in ethanolic, then a solution (solution α) of 0.55 mol/L was formed; on the other hand, a mixture of x Ni(OAc)<sub>2</sub> · 4H<sub>2</sub>O / (1 - x) Zn(OAc)<sub>2</sub> · 2H<sub>2</sub>O (where x = 0, 0.01, 0.02, and 0.05, respectively) were dissolved in DMSO to form another solution (solution β), for which the concentration was 0.10 mol/L. Then, under constant stirring, solution α was added drop wise into the solution β. Resultantly, a sol was formed and dispersed subsequently. After Ostwald ripening, the sol was iteratively washed by precipitation and resuspension in ethanol so as to remove the excess reactants (Ni<sup>2+</sup> and Zn<sup>2+</sup>). The resulting nanocrystals were then capped with dodecylamine. The resulting powders were then annealed at 500 °C for 30 min.

JEM-1010 transmission electron microscope (TEM) was used to study the sample morphology and size. The sample structure and composition were carried out by X-ray diffraction (XRD) using a D/MAX 2200 VPC X-ray diffractometer (Rigaku) with Cu-K<sub>α</sub> radiation of 1.54056 Å. Magnetic measurements were performed on a superconducting quantum interference device (SQUID) magnetometer (MPMS-XL7 Quantum Design, Inc).

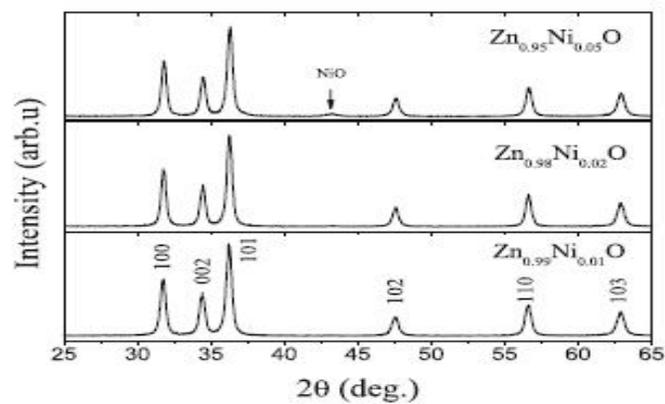
**Results and Discussion**

The sphere like morphology could be identified clearly by the TEM image (fig. 1) {TEM image of the sample Zn<sub>0.99</sub>Ni<sub>0.01</sub>O} of Ni-doped ZnO nanoparticles and average size of the nanoparticles is estimated to be about 25 nm.



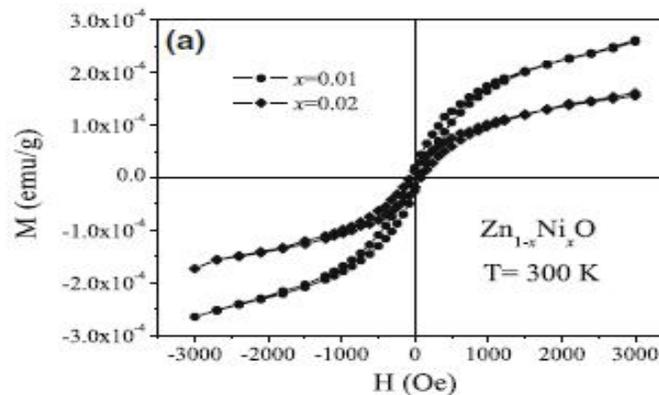
**Fig.1.** TEM image of the sample  $\text{Zn}_{0.99}\text{Ni}_{0.01}\text{O}$ .

XRD pattern (fig. 2) indicate that the synthesized sample are in wurtzite structure with all the peaks correspond to ZnO when the content of Ni ( $x$ ) is less than 0.05, however, when  $x$  reaches 0.05 an additional diffraction peaks observed due to phase segregation.

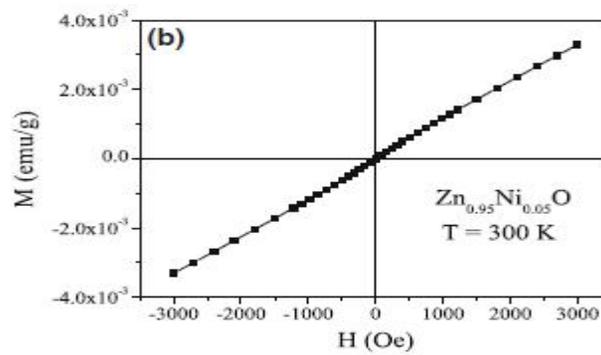


**Fig. 2.** XRD pattern of ZnNiO nanoparticles.

SQUID results (fig. 3) indicate weak magnetic hysteresis loop when the content Ni ( $x$ ) is less than 0.05 while at  $x=0.05$  particles show paramagnetism (fig.4).



**Fig. 3.** M-H curves of  $\text{Zn}_{1-x}\text{Ni}_x\text{O}$  nanoparticles ( $x \leq 0.02$ ).



**Fig. 3.** M-H curve of  $Zn_{1-x}Ni_xO$  nanoparticles ( $x \leq 0.05$ ).

### Conclusion

In conclusion, Ni-doped ZnO ( $Zn_{1-x}Ni_xO$ ,  $0 \leq x \leq 0.05$ ) diluted magnetic semiconductors nanoparticles are prepared by sol-gel process. XRD analysis shows wurtzite structures for all the  $Zn_{1-x}Ni_xO$  nanoparticles. With increasing Ni concentration, the wurtzite structures degrade gradually. FM is observed in  $Zn_{1-x}Ni_xO$  ( $x \leq 0.02$ ) system, and the Curie temperature is higher than 300 K. It is probable that the RT FM might originate from long-range  $Ni^{2+}$ - $Ni^{2+}$  ferromagnetic coupling mediated by shallow donor electrons. However, when x increasing to 0.05, antiferromagnetic NiO formed and the system turns to be paramagnetic.

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